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Example: The first box is the intersection between the zinc cation and the chloride anion, so you should write ZnCl2, as shown. zinc iron (II) iron (III) gallium silver lead (IV) chloride ZnCl2 acetate nitrate oxide nitride sulfate Write the formulas for the following compounds: 1) copper (II) chloride _____ 2) lithium acetate _____ 3) vanadium (III) selenide _____ 4) manganese (IV) nitride _____ 5) beryllium oxide _____ 6) sodium sulfate _____ 7) aluminum arsenide _____ 8) potassium permanganate _____ 9) chromium (VI) cyanide _____ 10) tin (II) sulfite _____ 11) vanadium (V) fluoride _____ 12) ammonium nitrate _____ Ionic Compound Naming Chilton Honors Chemistry Chemical Formula Writing Worksheet Solutions Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. 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